

# Management of Criticality in Subcritical Solution Systems

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# Reactivity Effects in Solution Systems

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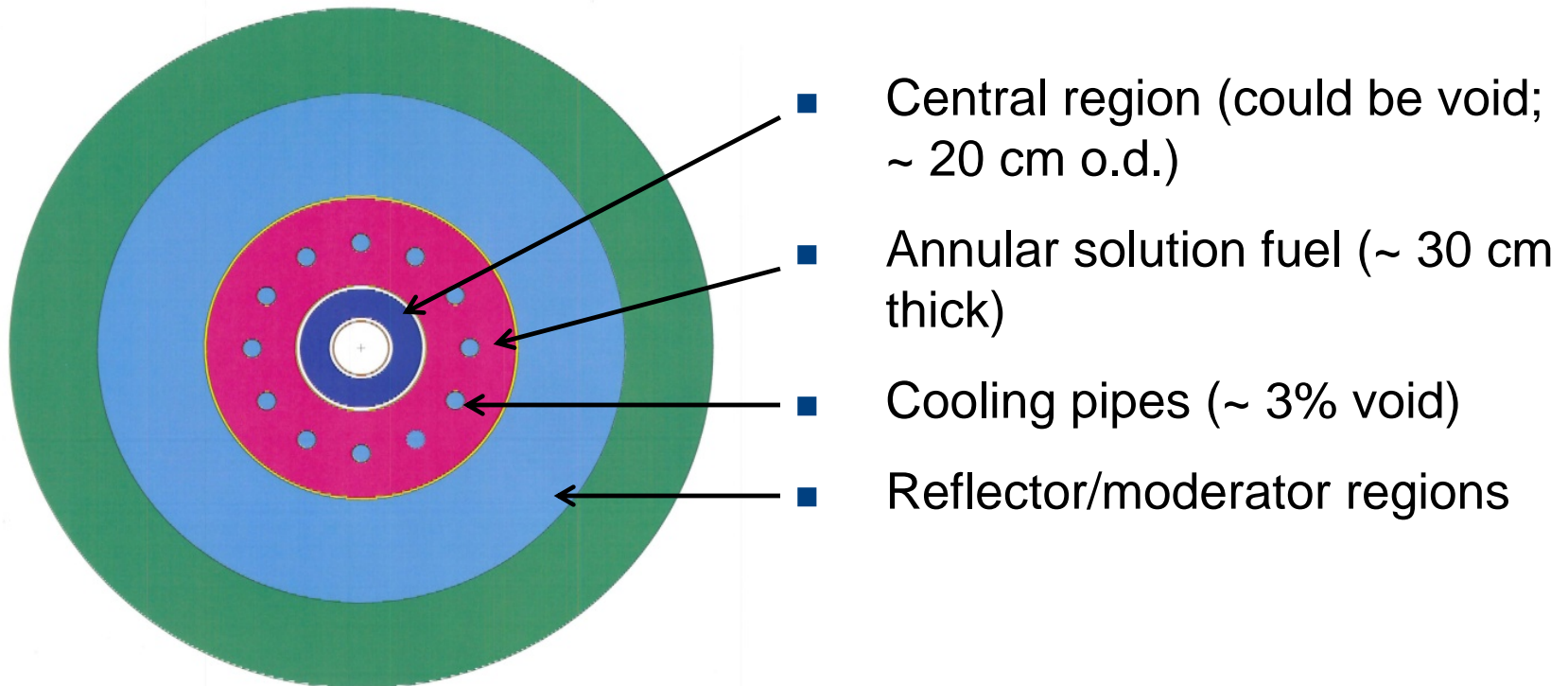
- Short Term (~ minutes)
  - Fission induced heating causes solution temperature to rise resulting in lower density and increased volume (reactivity drops)
  - Radiolytic gas formation induces void in solution that increases from solution bottom to surface causing drop in average density and solution flow; these effects combined cause rapid reactivity fluctuation around the average value
- Mid Term (~ hours to days)
  - Hydrogen loss due to water radiolysis results in reactivity decrease
  - Nitrogen loss in uranyl nitrate systems results in reactivity increase
- Long Term (~ weeks to months)
  - $^{235}\text{U}$  burn-up results in lower reactivity (10 – 20% over one year)
  - Fission product inventory build up results in lower reactivity (10 – 20% over one year)

# Subcritical System Design Process

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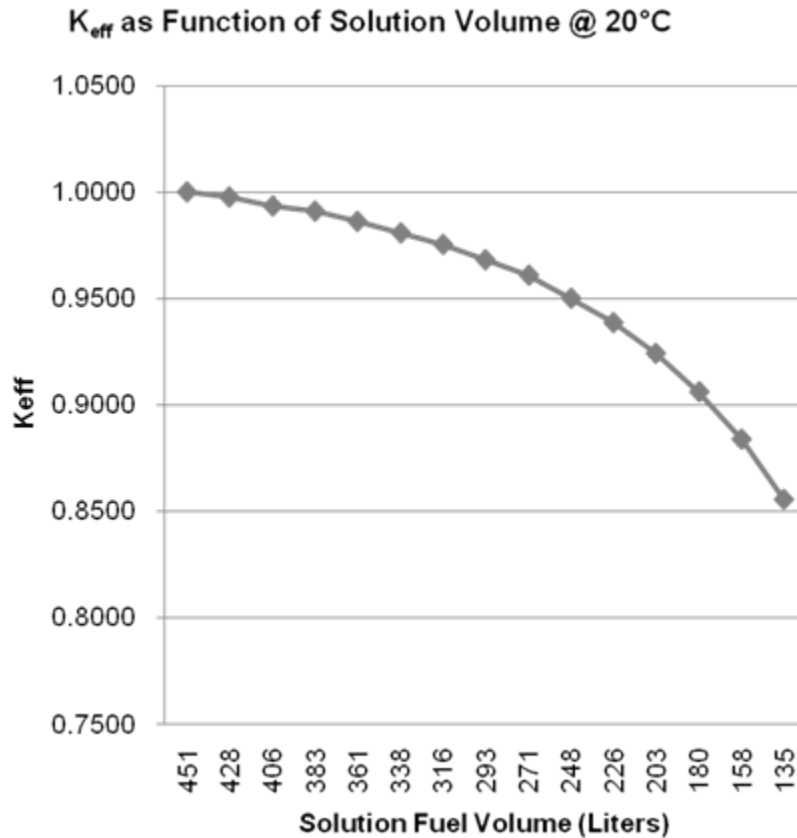
- Object is to maximize fissions per source neutron (direct measure of  $^{99}\text{Mo}$  production)
- Select candidate Low Enriched Uranium (LEU) fuel
  - Uranium concentration (gU/liter)
  - Solvent (heavy or light water)
- Select candidate vessel configuration
- Estimate critical fuel volume for candidate fuel and vessel configuration
- Reduce fuel volume by 5%
- Estimate production for reaction vessel at 20° C, 60° C (operating design point) and 80° C
- Estimate production effects of variation of reactivity over time

# Subcritical Solution Fueled System Configuration



- 90 gU/liter (19% enriched) LEU
- 451 liters full volume

# Keff as Function of Solution Volume

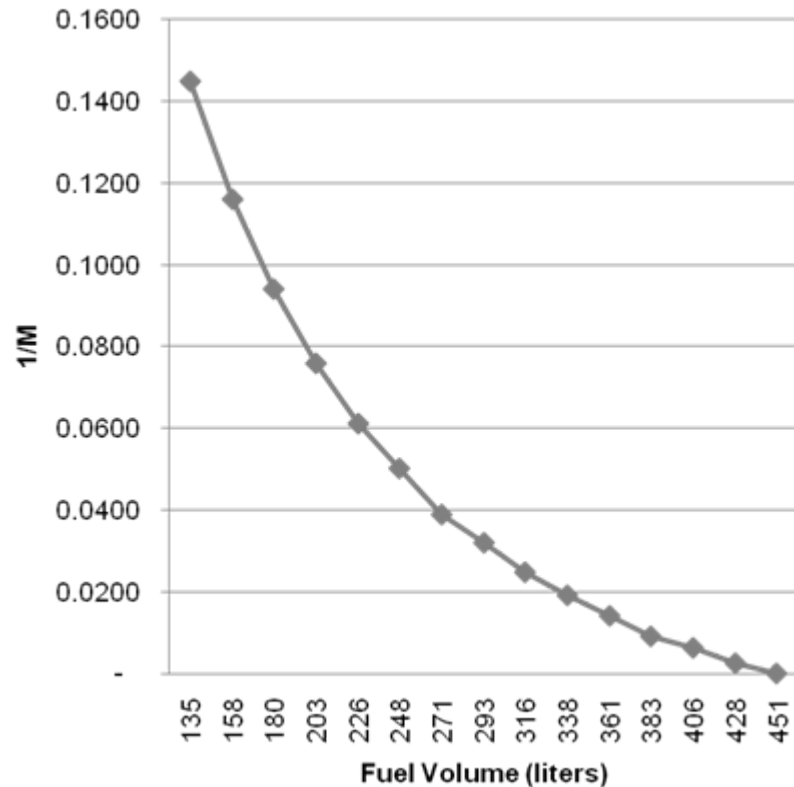


Volume (liters)	Temperature (° C)	Keff
451	20	1.0000
428 (-5%)	20	0.9974
250	20	0.9500
451	60	0.9930

***Note that heating fuel at full volume is equivalent in  $k_{eff}$  to 50 liter reduction from critical volume***

# Startup & Operation of Subcritical Solution Systems

## 1/M Approach to Critical



## Comments

- Curve is conservative throughout (always predicts less volume than actual)
- Even when physically far from critical  $k_{\text{eff}}$  is above 0.95
- Accurate estimate of critical volume not obtained until above 95% of actual value
- Volume control of < 0.5 liter is relatively straightforward.

# Conclusions

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- For the class of configuration of fissile solution systems typical of that considered for subcritical production operations  $k_{\text{eff}}$  is not a useful measure of the margin from criticality
- A physical measure such as solution volume is an accurate reference for stand-off from critical
- Nature of fissile solution systems is that once fission heating and radiolytic gas production starts the reactivity drops significantly.
- Fissile solution systems may be operated reliably and safely with  $k_{\text{eff}}$  approaching delayed critical.